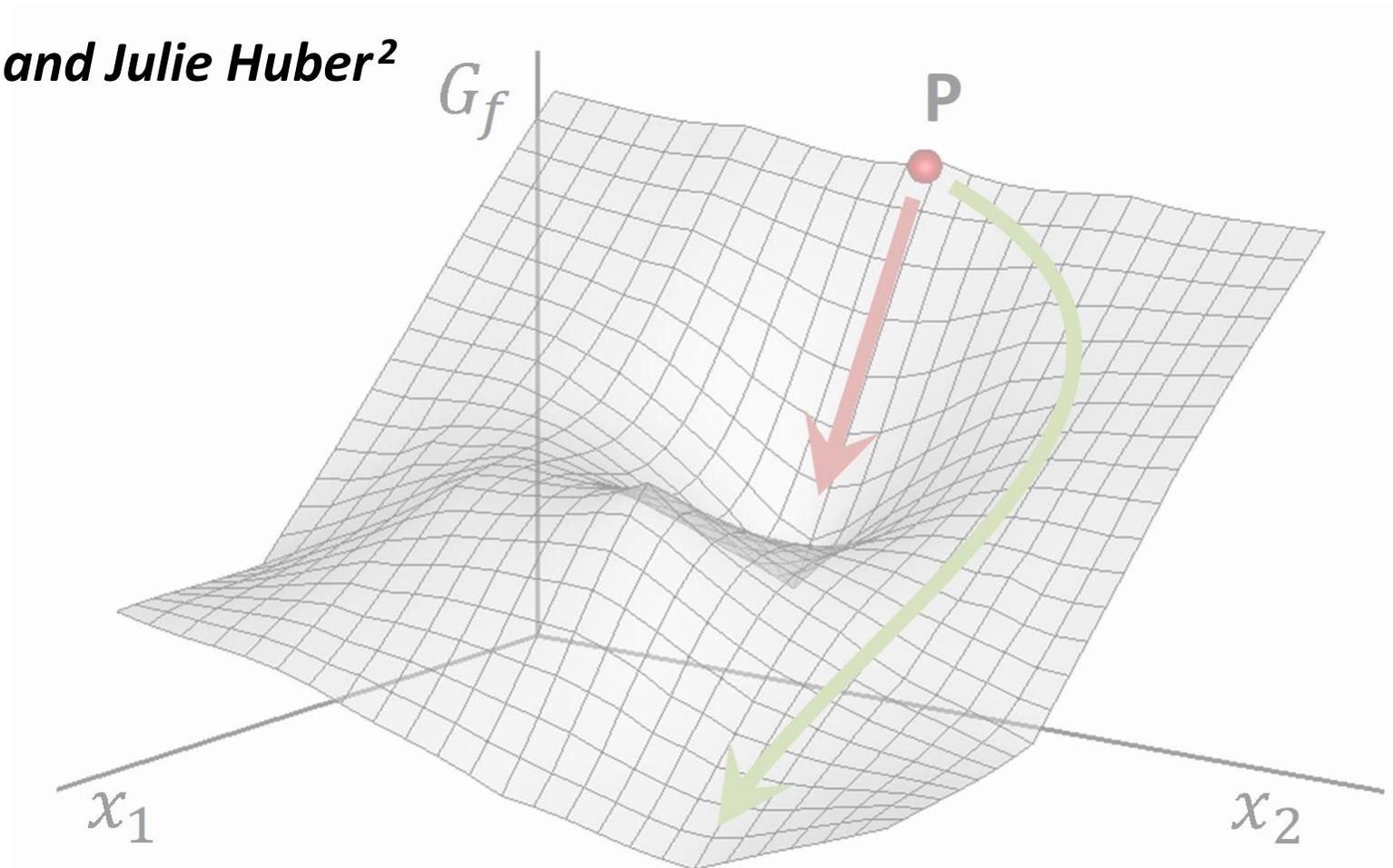


Using the maximum entropy production principle to understand and predict microbial biogeochemistry

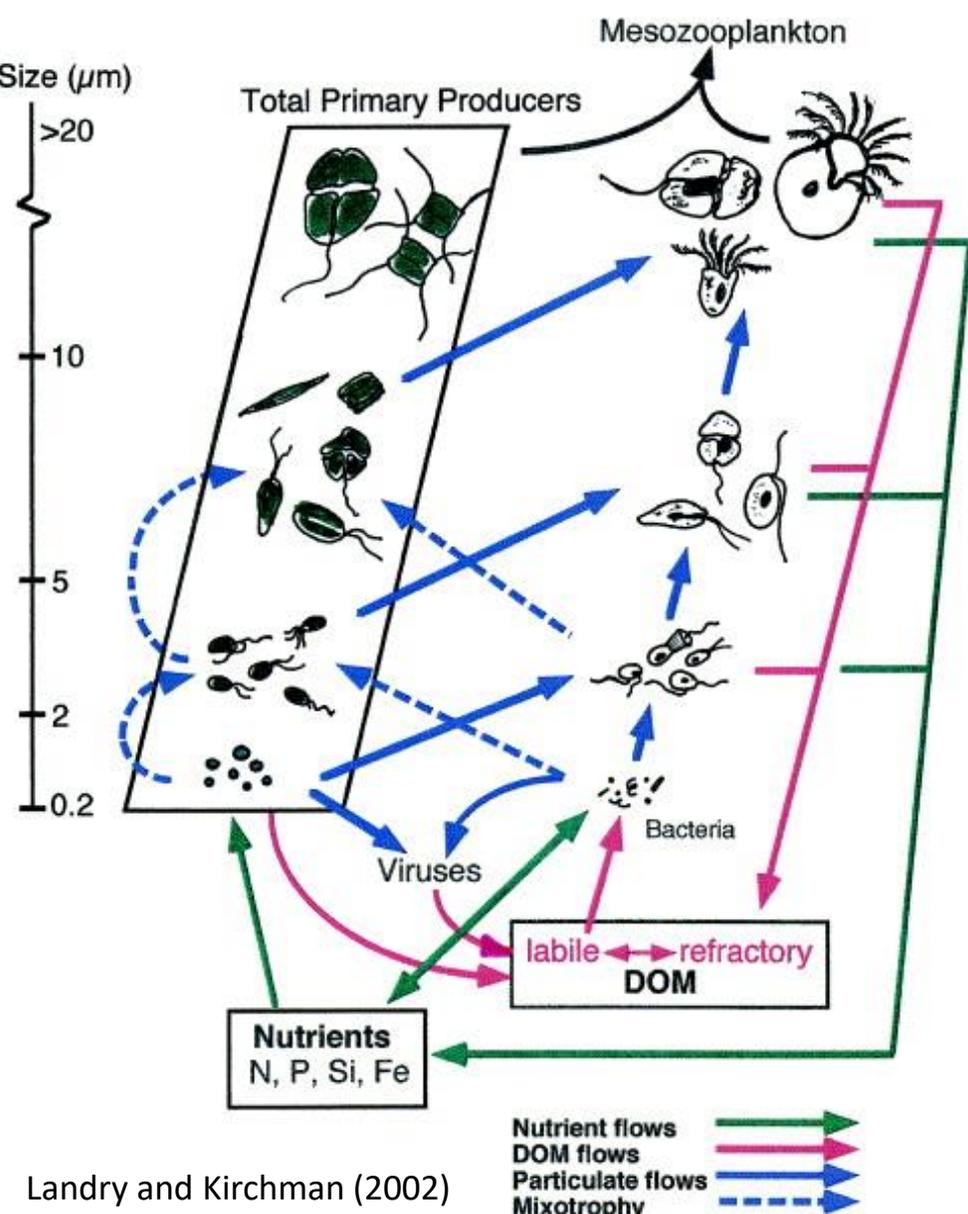
Joe Vallino¹, Ioannis Tsakalakis¹, and Julie Huber²

¹ Ecosystems Center
Marine Biological Laboratory
Woods Hole

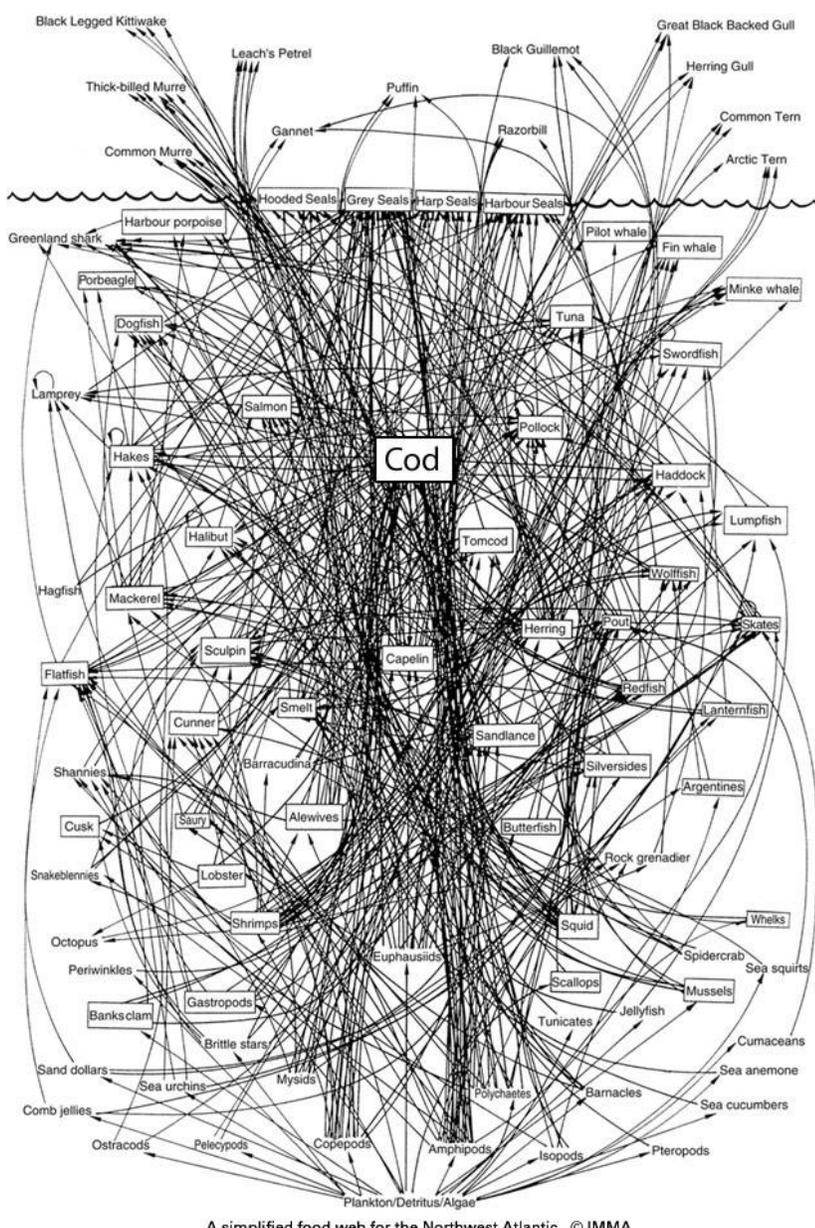
² Marine Chemistry and Geochemistry
Woods Hole Oceanographic Institution



Reductionist Modeling Approach, Marine Biogeochemistry

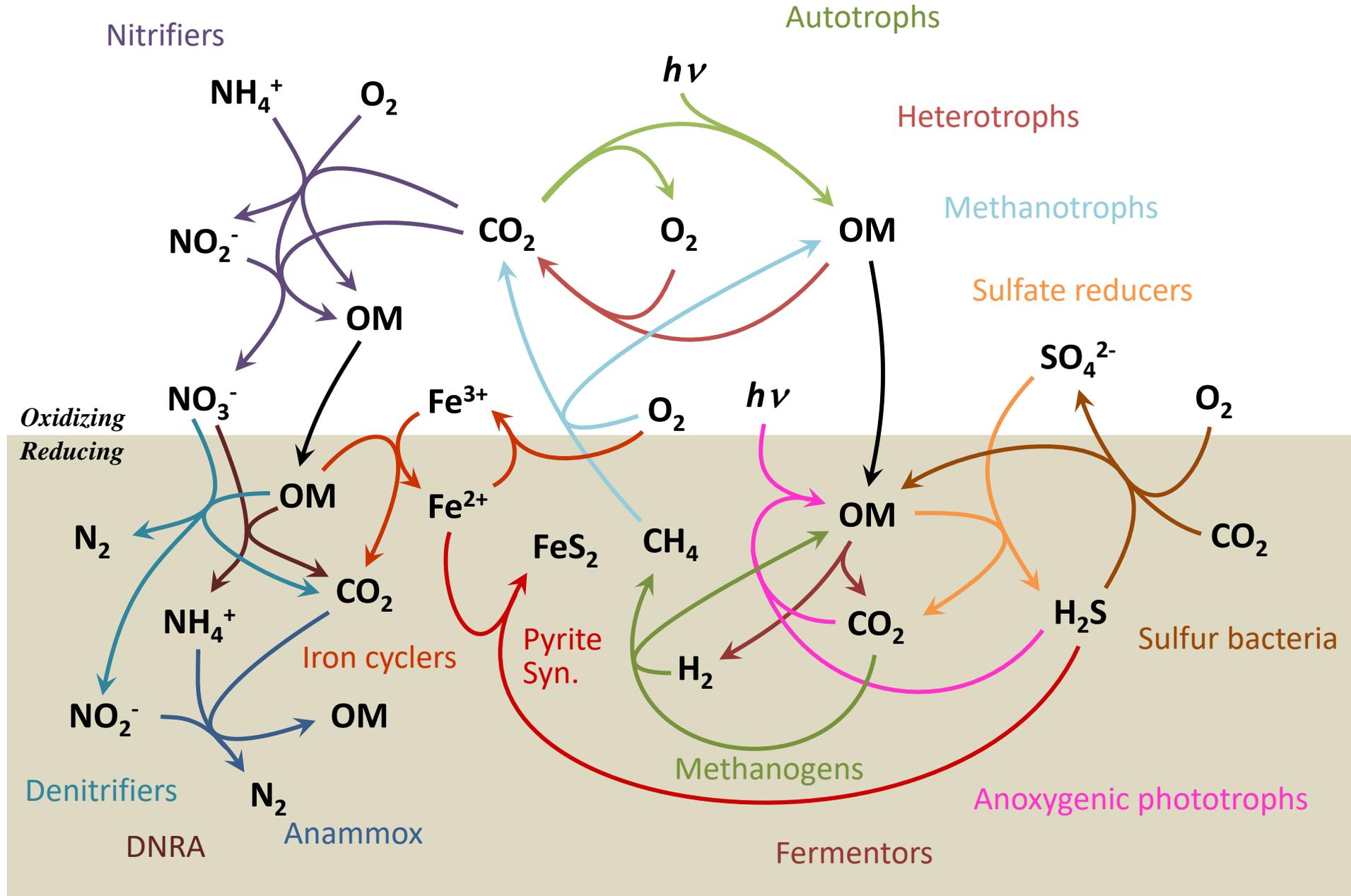


Landry and Kirchman (2002)



- Aggregation causes errors
- Huge number of parameters typically
- Insufficient information
- What happens when the community composition changes?
- How about the Rare Biosphere?

Focus on functions that dissipate energy instead



Pathways are distributed across phyla

No centralized control

Which pathways are up-regulated?

Which are down-regulated?

How do resources limit?

Many degrees of freedom

Does "who's there" matter?

How to predict protein allocation?

Use optimization-base approach!

Maximum Entropy Production (MEP)

Steady state nonequilibrium systems with many degrees of freedom will likely organize to maximize the rate of entropy production.

Examples: Fire, Hurricanes, and Living systems.

See:
Ziegler 1963
Paltridge 1975
Dewar 2003, 2005
Martyushev & Seleznev 2006

Entropy?

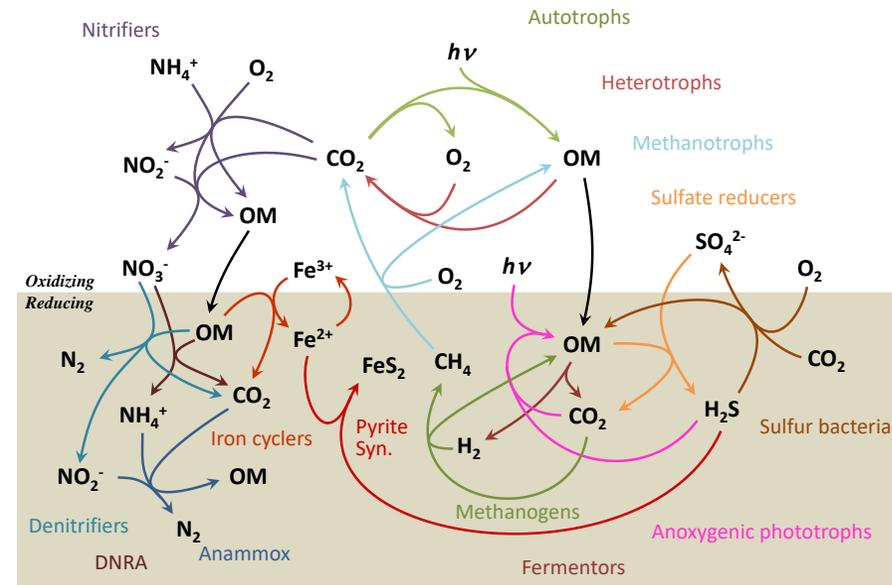
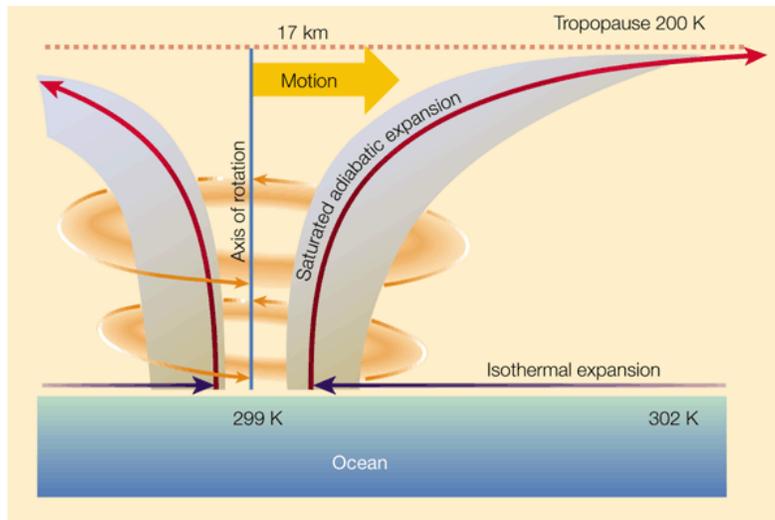
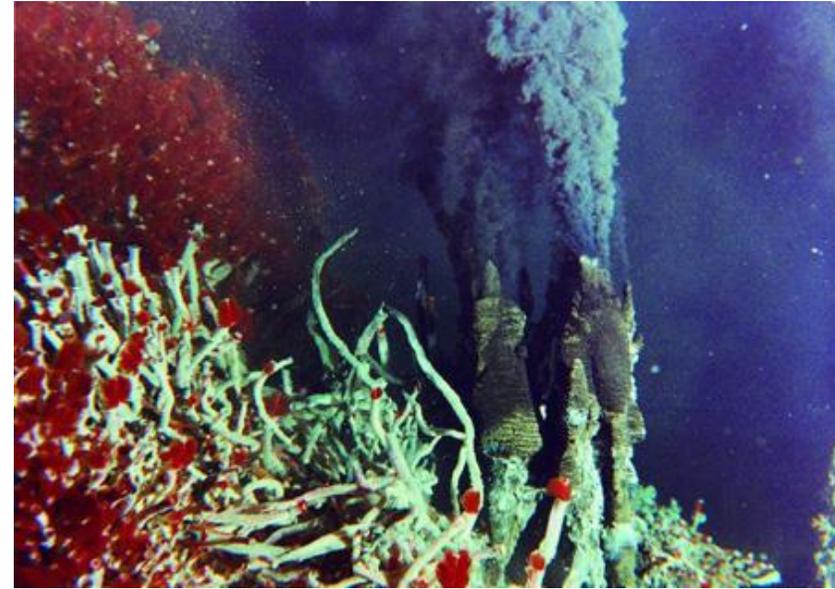
- Dispersal (spreading out) of energy
- Order is usually a small and unimportant component of entropy
- Order must contend with Boltzmann's constant: $k=1.3806488 \times 10^{-23} \text{ J K}^{-1}$
- Living organisms are not low entropy structures.

Easier to think about entropy as the destruction of *free* energy†

† Energy is conserved, free energy (or exergy) is not.

Maximum Entropy Production (MEP)

Or: Systems organize to maximize dissipation rate of energy potentials

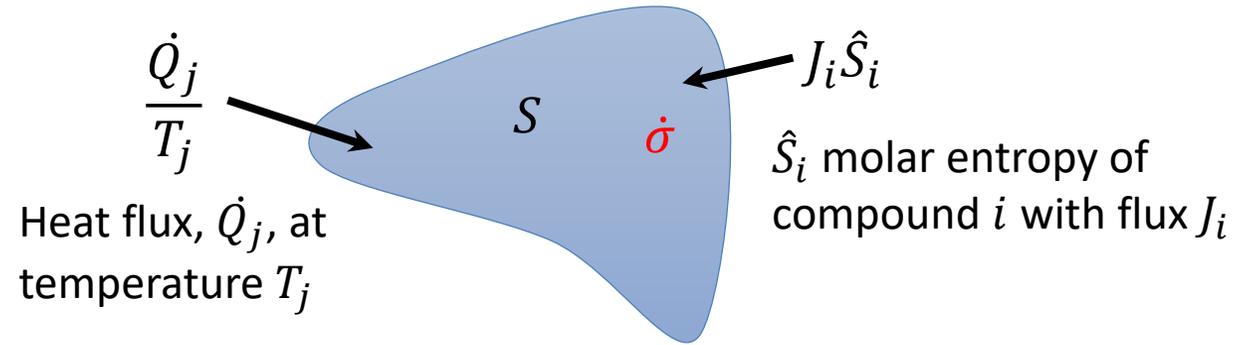


MEP, Which Entropy Term?

Entropy Balance Equation:

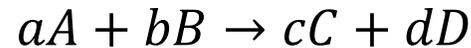
$$\frac{dS}{dt} = \sum_i J_i \hat{S}_i + \sum_j \frac{\dot{Q}_j}{T_j} + \dot{\sigma}$$

where $\dot{\sigma}$ is internal entropy production from irreversible processes. Second Law says $\dot{\sigma} \geq 0$



MEP concerns $\dot{\sigma}$ not S

For Chemical Reaction



Gibbs free energy

$$\Delta_r G(T, P) = \Delta_r G^\circ(T, P) + RT \log \left(\frac{(\gamma_c [C])^c (\gamma_d [D])^d}{(\gamma_a [A])^a (\gamma_b [B])^b} \right)$$

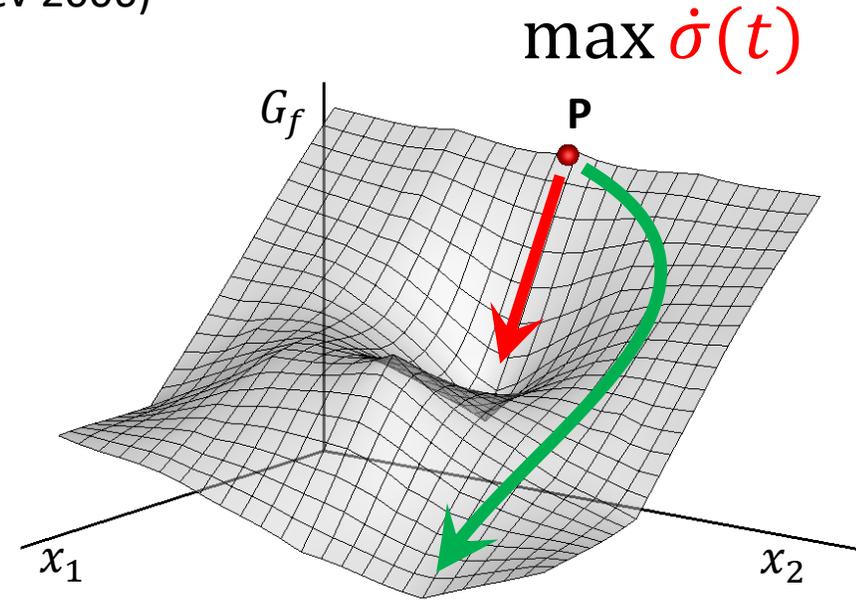
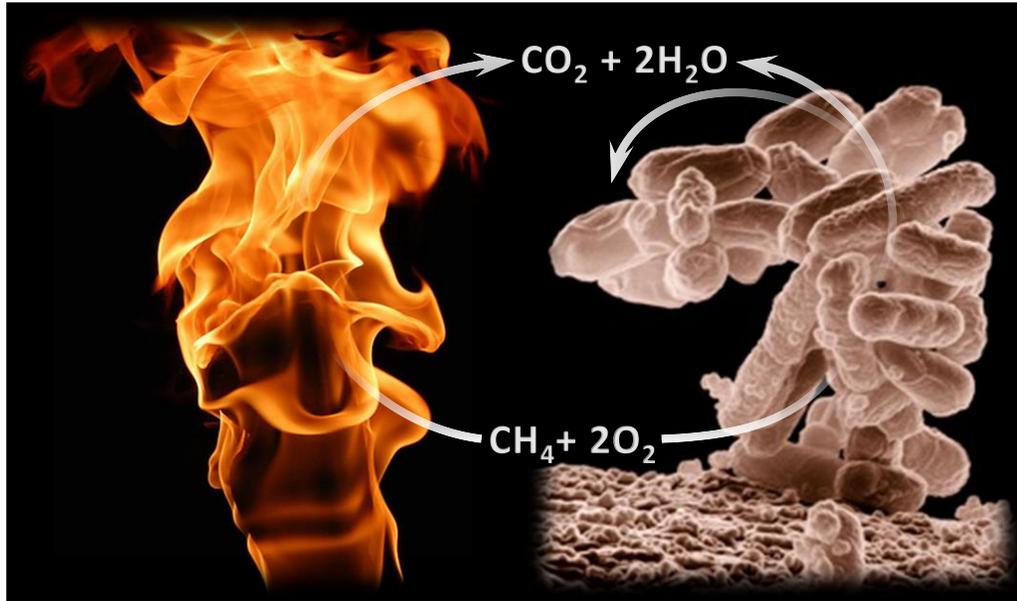


If free energy is not stored, then:

$$\dot{\sigma} = -\frac{V}{T} r \Delta_r G(T, P)$$

Maximum Entropy Production for non-steady state systems

MEP only derived for SS systems (Dewar 2003, Martyushev & Seleznev 2006)



Difference between abiotic (fire) and Life:

- **Abiotic:** maximizes instantaneous $\dot{\sigma}(t)$
- **Life:** maximizes time-averaged $\langle \dot{\sigma}(t) \rangle$

Temporal strategies are a hallmark of biology:

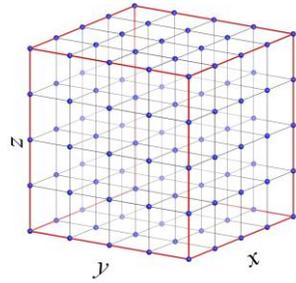
circadian rhythm, resource storage, dormancy, life cycles, anticipatory control, etc.

$$\max \langle \dot{\sigma}(t) \rangle = \max \left(\frac{1}{\Delta t} \int_t^{t+\Delta t} \dot{\sigma}(\tau) d\tau \right)$$

$$\max \langle \dot{\sigma}(t) \rangle \geq \frac{1}{\Delta t} \int_t^{t+\Delta t} \max \dot{\sigma}(\tau) d\tau$$

Coordination Over Space

Entropy production can also be increased with coordination over space (see Vallino 2010)



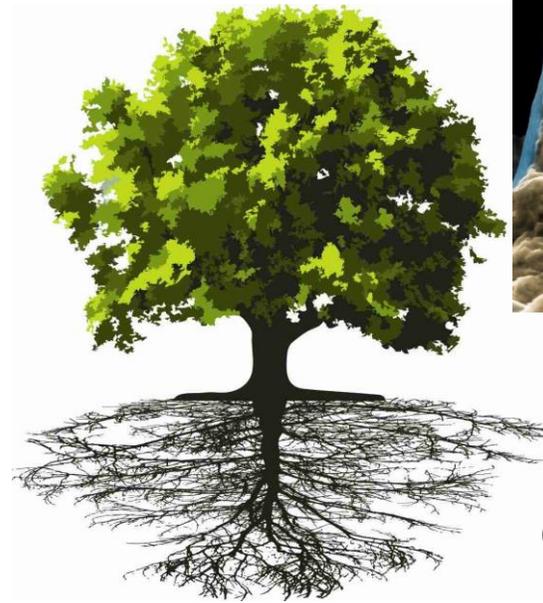
$$\langle \dot{\sigma} \rangle_{\Omega} = \max \frac{1}{V} \iiint_{\Omega} \dot{\sigma}(x, y, z) d\Omega$$

$$\langle \dot{\sigma} \rangle_{\Omega} \geq \frac{1}{V} \iiint_{\Omega} \max \dot{\sigma}(x, y, z) d\Omega$$

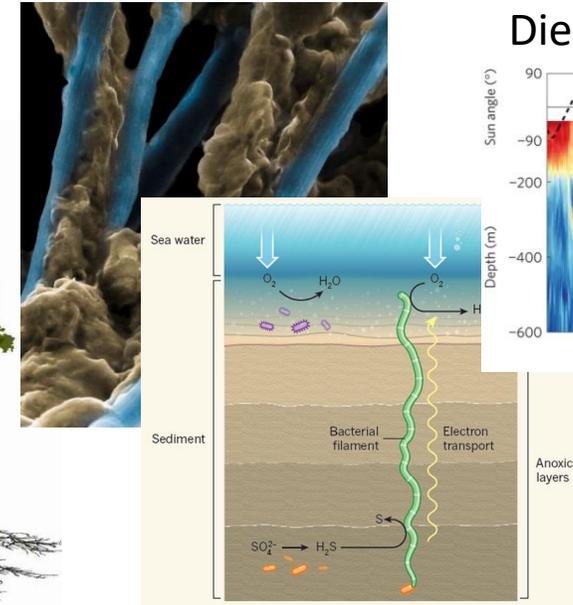
Living systems evolve to maximize energy dissipation over the greatest possible spatial and temporal scales

“Paradigm shift, from **‘we eat food’** to **‘food has produced us to eat it’**” (Lineaweaver&Egan 2008)

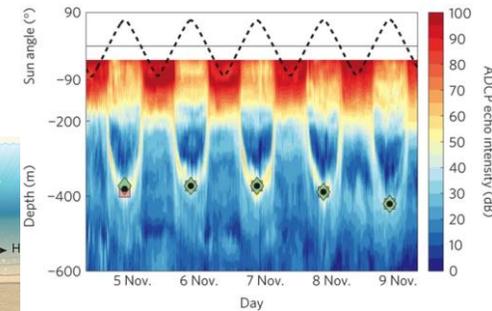
Multicellularity



Cable bacteria



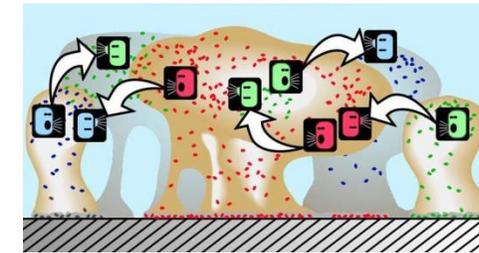
Diel vertical migration



Stigmergy



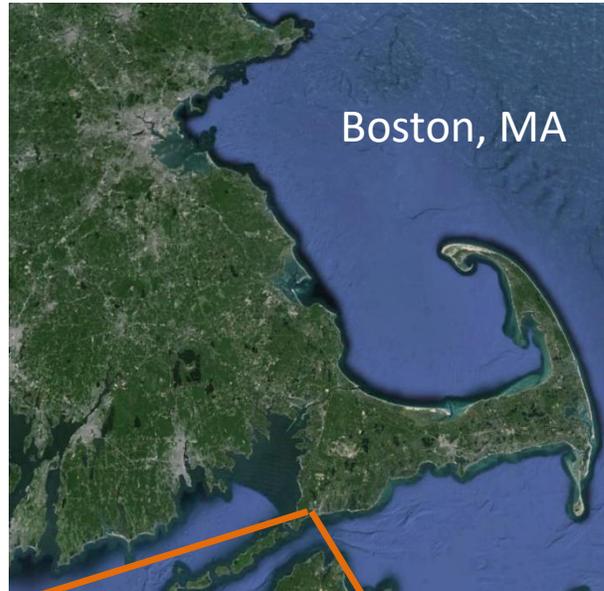
Quorum sensing



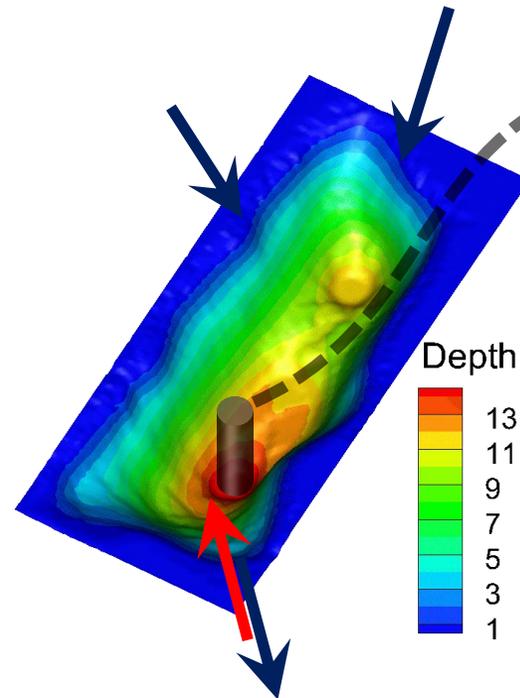
MEP approach to modeling biogeochemistry

- Represent biogeochemistry as a distributed metabolic network focused on redox reactions
- Allocate catalysts (protein) to metabolic pathways that maximize entropy production **over time and space**
- Optimization replaces need to understand how communities assemble (aka, climate verses weather modeling)
- Replace parameters with optimization variables

Example: Siders Pond “Laboratory”

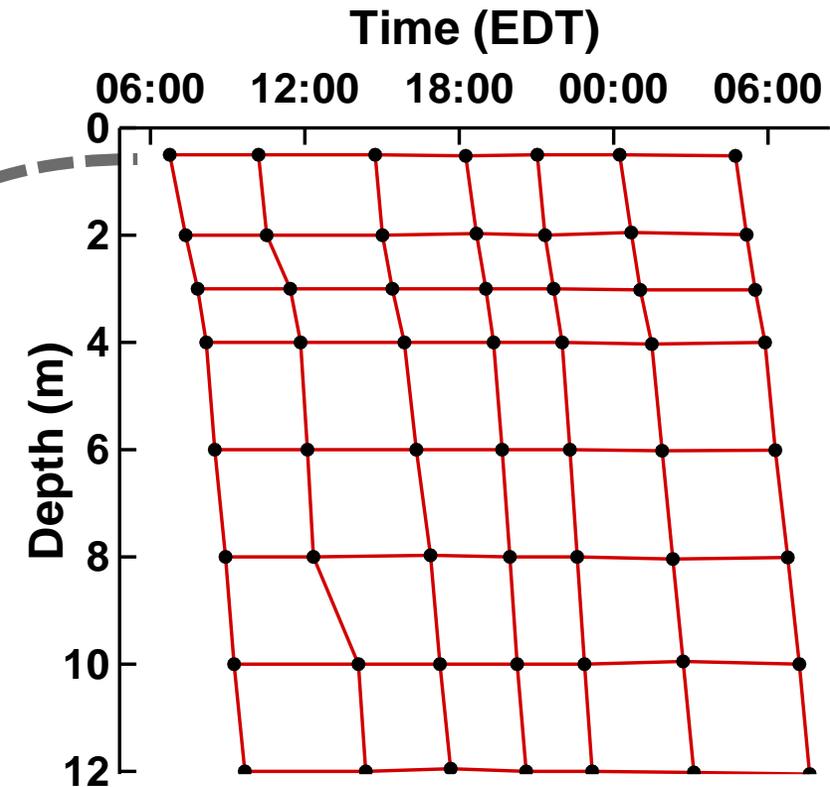


Meromictic Coastal Pond
Permanently stratified
Depth: 15 m max
Area: 14 hectares
Volume: $\sim 1 \text{ Mm}^3$

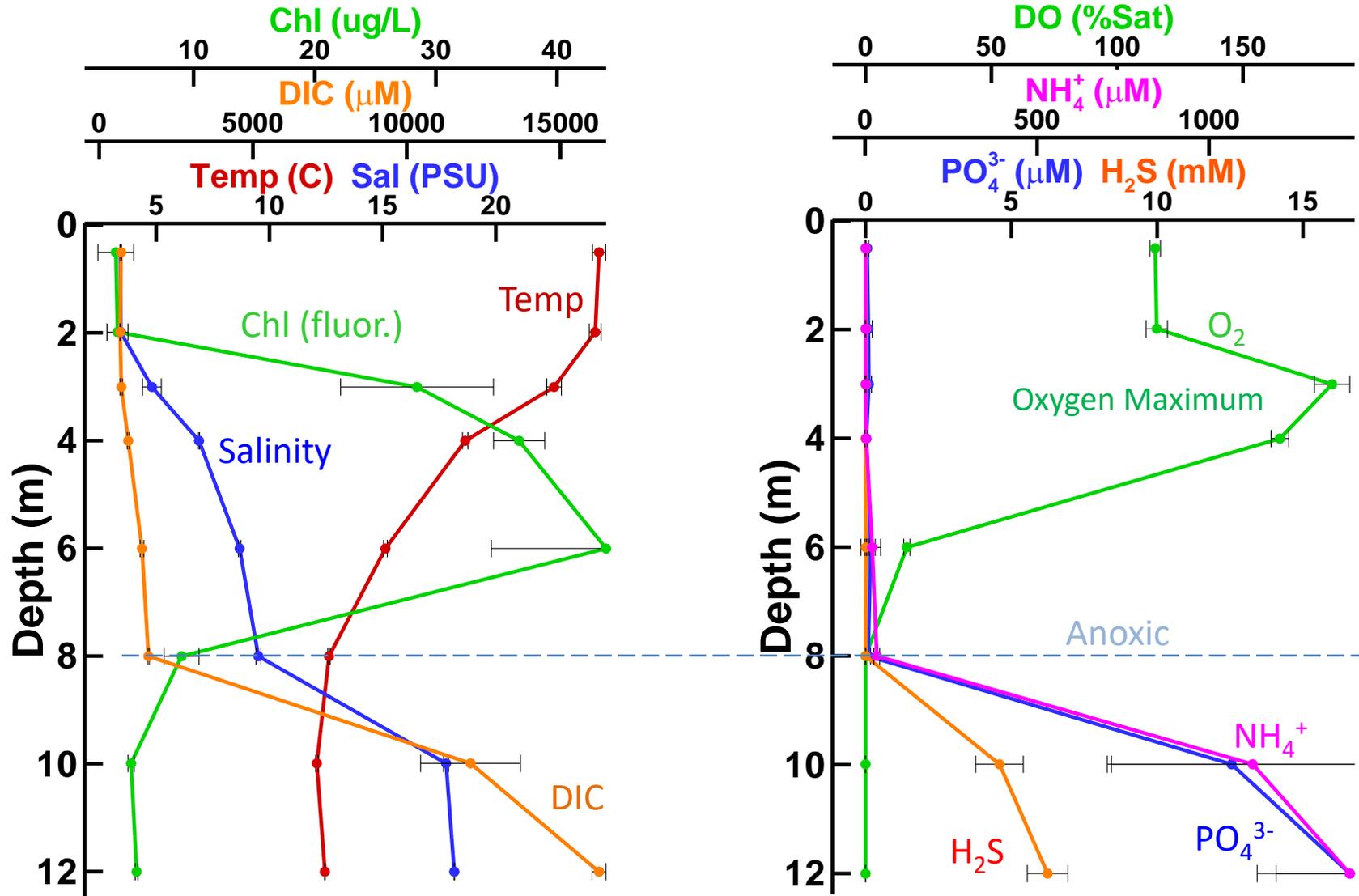


Space-Time Sampling Grid

- Biogeochemistry
- Metagenomics (cast 1)
- Metatranscriptomics (all casts)



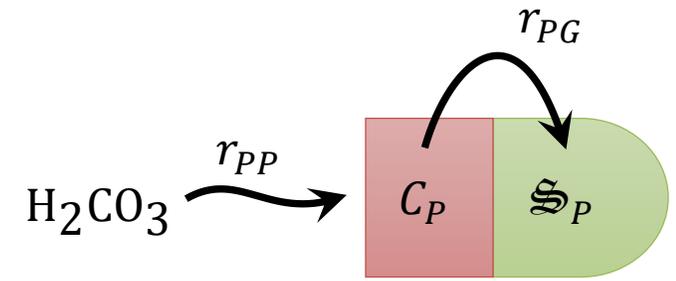
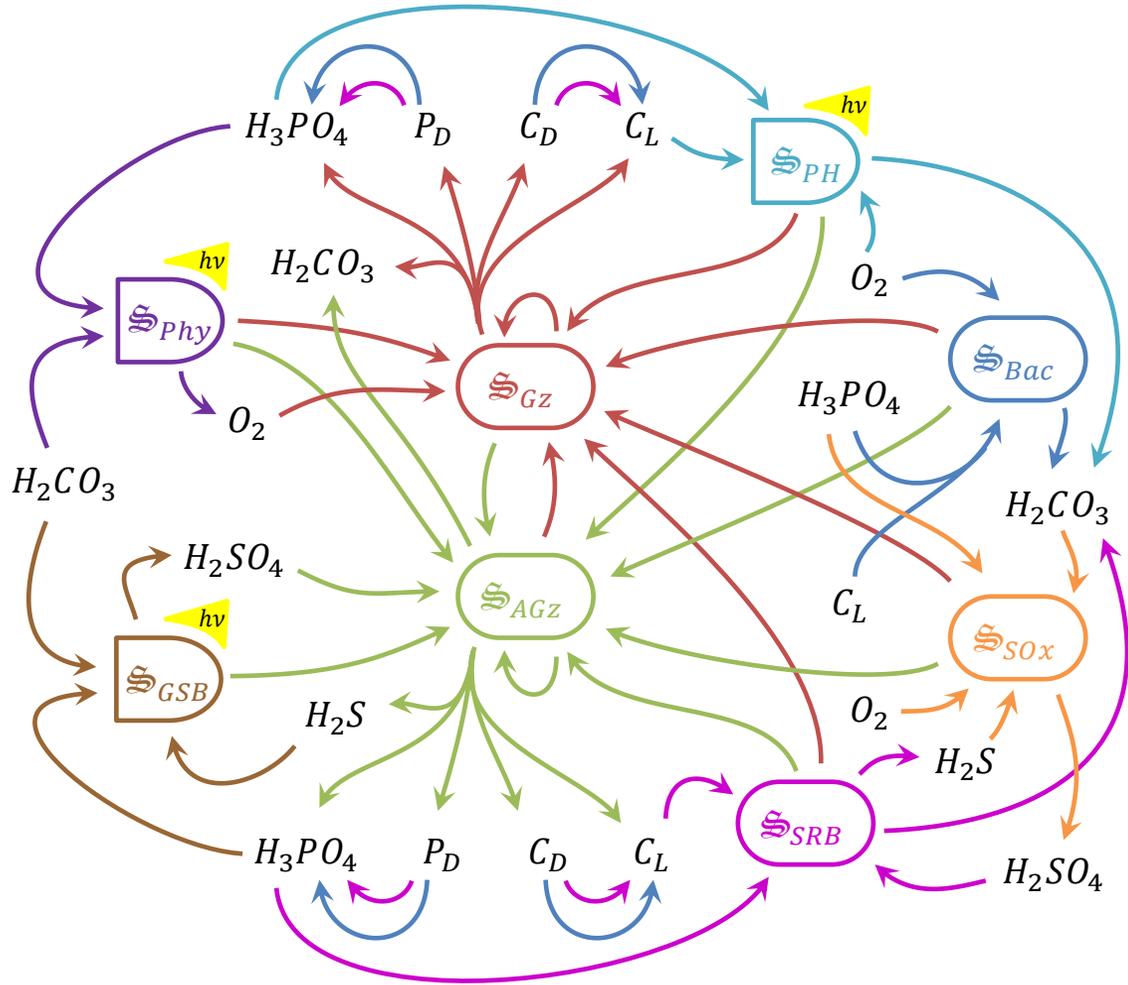
Vertical Gradients



Large spatial gradients (over *m* not *mm*)

Distributed metabolic network

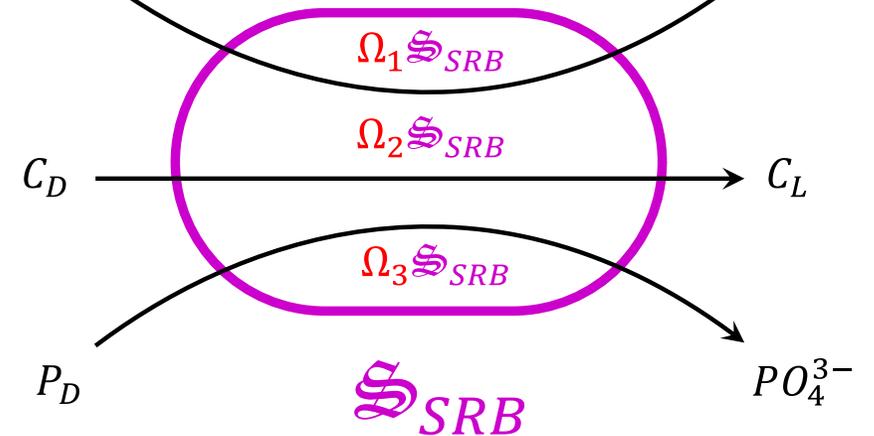
Vallino and Huber 2018



$$r_{PP}: \epsilon_P H_2CO_3 + n_\gamma h\nu \xrightarrow{\Omega_{1,P} S_P} \epsilon_P (C_P + O_2)$$

$$r_{PG}: C_P + (1 - \epsilon_P) O_2 + \epsilon_P \delta H_3PO_4 + \dots \xrightarrow{\Omega_{2,P} S_P} \epsilon_P S_P + (1 - \epsilon_P) H_2CO_3$$

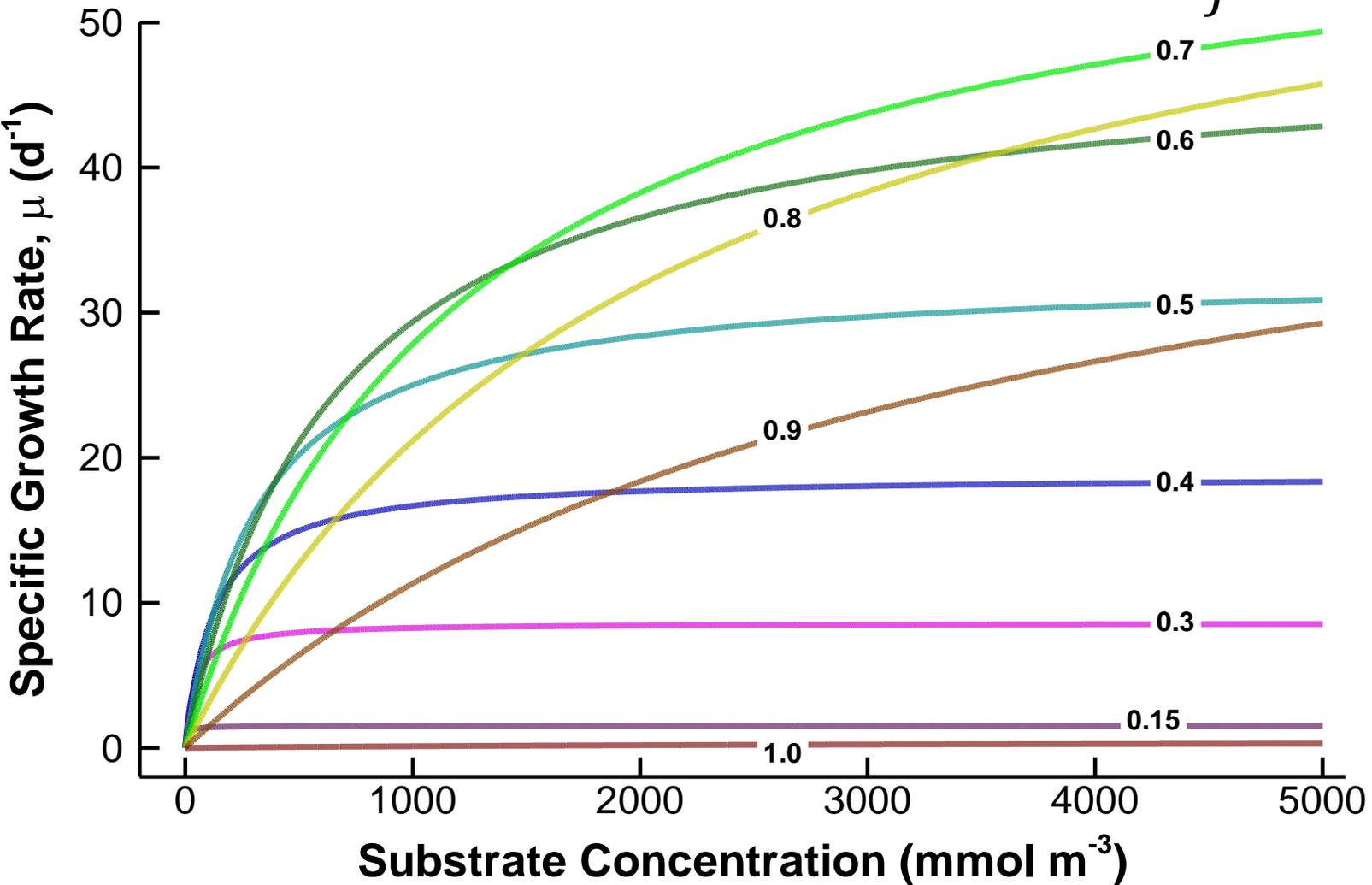
$$C_L + \epsilon_{SRB} (\gamma NH_4^+ + \delta PO_4^{3-}) + \frac{1}{2} (1 - \epsilon_{SRB}) SO_4^{2-} \xrightarrow{\epsilon_{SRB} S_{SRB} + (1 - \epsilon_{SRB}) \times (\frac{1}{2} H_2S + CO_2)}$$



Reaction Kinetics (adaptive Monod)

$$r_{i,j} = v^* \epsilon_j^2 \prod_{k=1}^{n_c} \left(\frac{C_k}{C_k + \kappa^* \epsilon_j^4} \right)^{\Lambda_{i,j,k}} \Omega_{i,j} C_{S_j} F_T(\epsilon_j, \Omega_{i,j})$$

Curves parameterized by ϵ_j



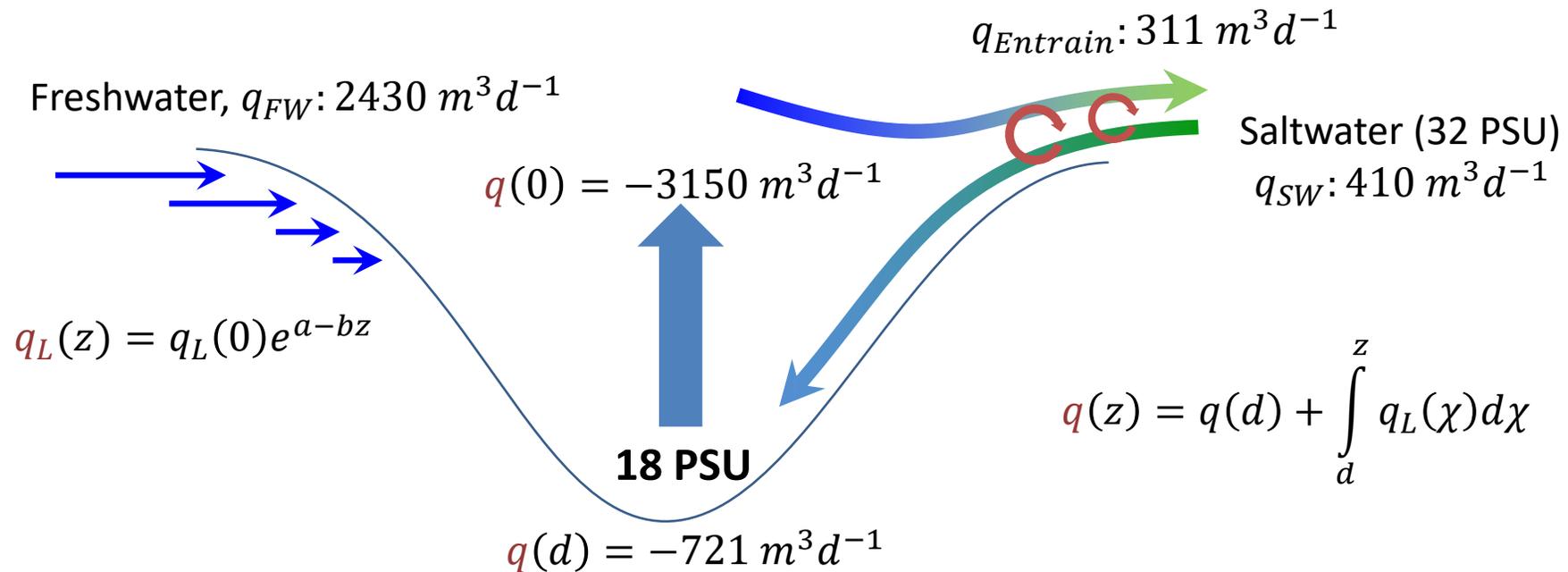
Siders Pond Transport (1D approximation)

$$\frac{\partial \mathbf{c}(t, z)}{\partial t} = D(z) \frac{\partial^2 \mathbf{c}(t, z)}{\partial z^2} + \left(\frac{D(z) \partial A(z)}{A(z) \partial z} + \frac{\partial D(z)}{\partial z} - \frac{q(z)}{A(z)} \right) \frac{\partial \mathbf{c}(t, z)}{\partial z} - \frac{\mathbf{c}(t, z) \partial q(z)}{A(z) \partial z} + \frac{q_L(z) \mathbf{c}_L(z)}{A(z)} + \mathbf{r}(t, z, \boldsymbol{\varepsilon}, \boldsymbol{\Omega}),$$

$$\text{BC: } \left. \frac{\partial \mathbf{c}(t, z)}{\partial z} \right|_{z=0} = 0 \quad \left(-D(z)A(z) \frac{\partial \mathbf{c}(t, z)}{\partial z} + q(z)\mathbf{c}(t, z) \right)_{z=d} = q(d)\mathbf{c}_B$$

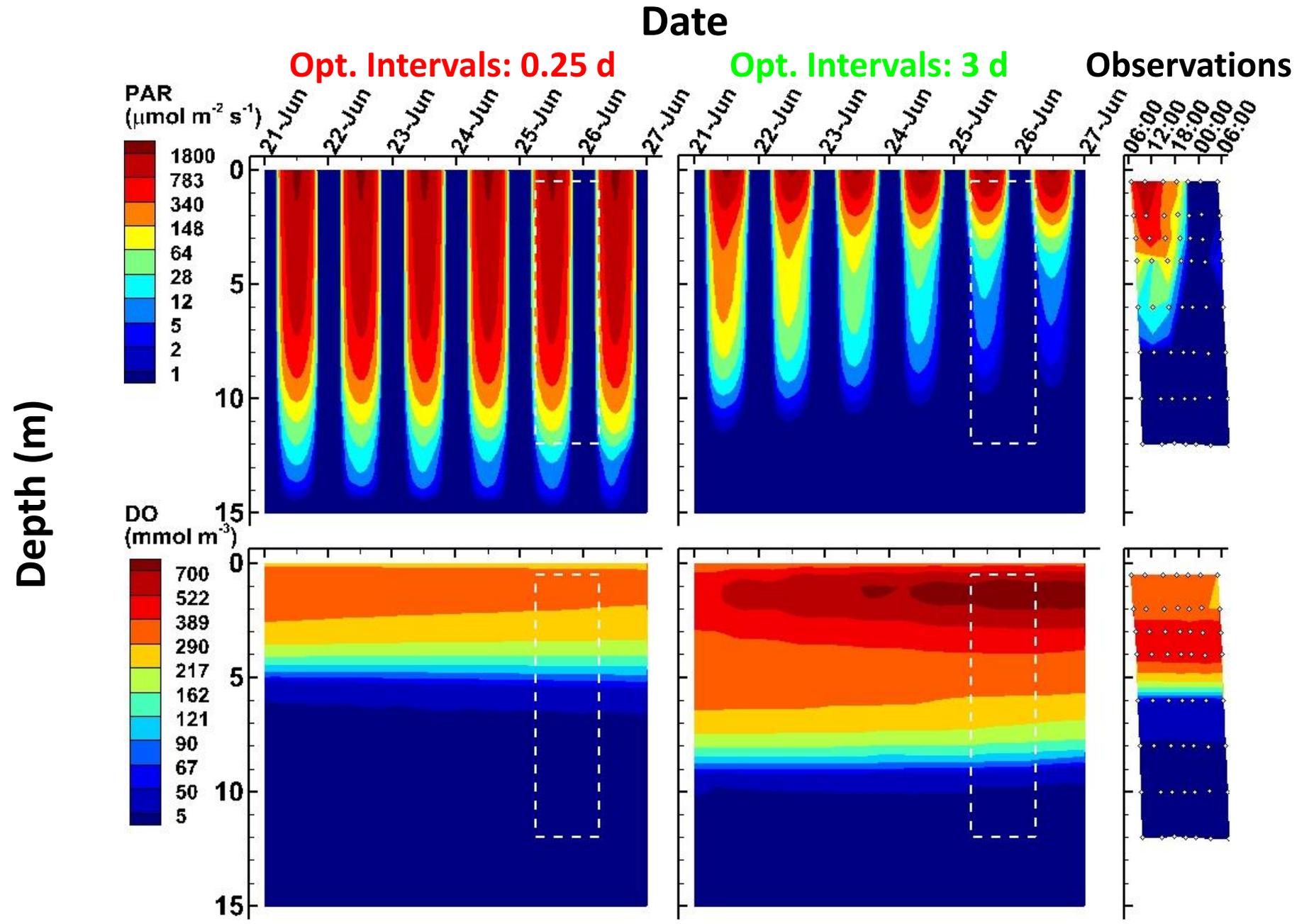
Dispersion coefficient, $D(z)$, was determined by fitting predicted to observed salinity vertical profile

Volumetric flow: $q(z)$ and lateral inputs: $q_L(z)$, $\mathbf{c}_L(z)$ obtained from observations and assuming:

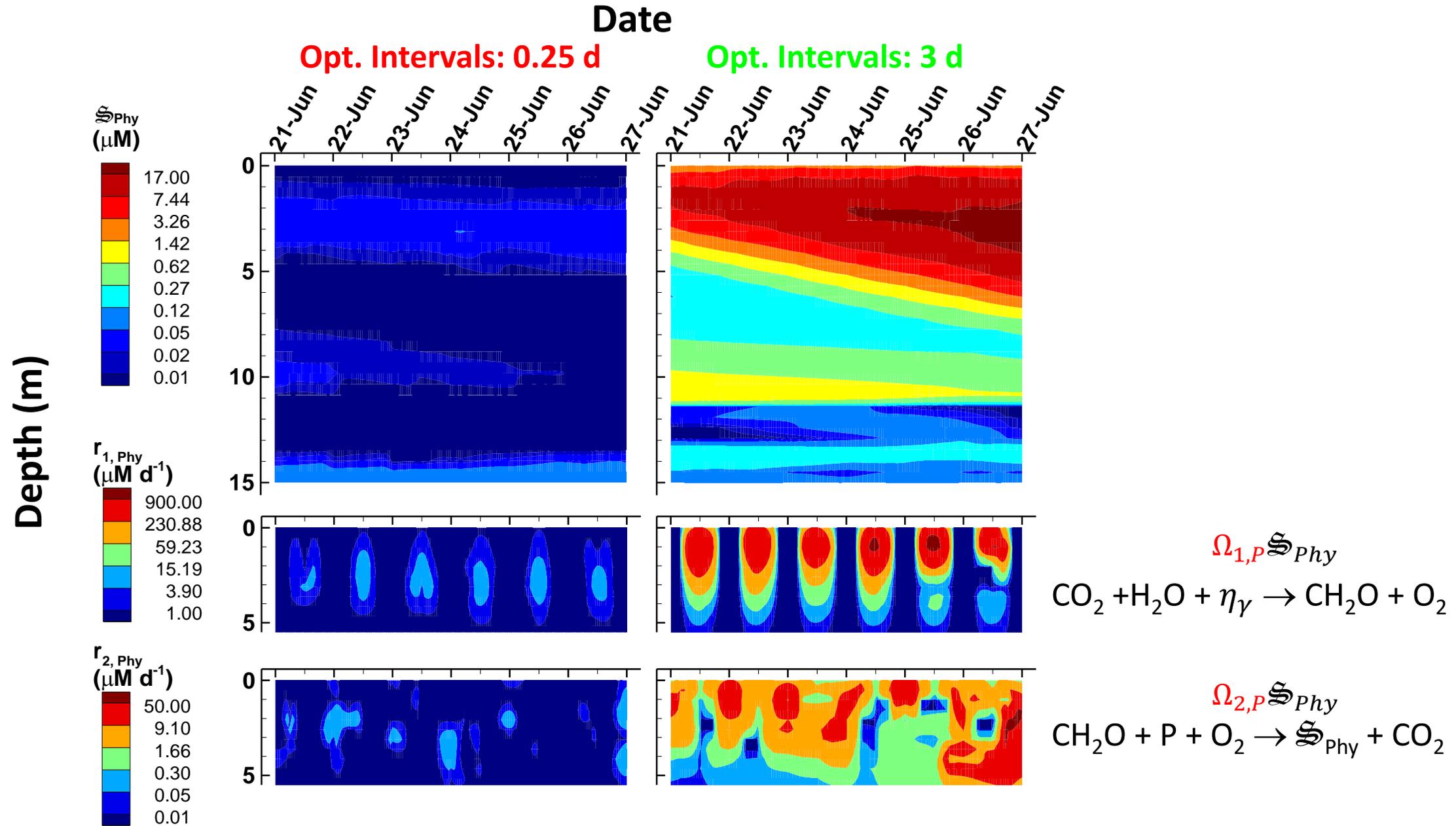


Short (0.25 d) vs Long (3 d) optimization: PAR & DO

Vallino & Huber 2018



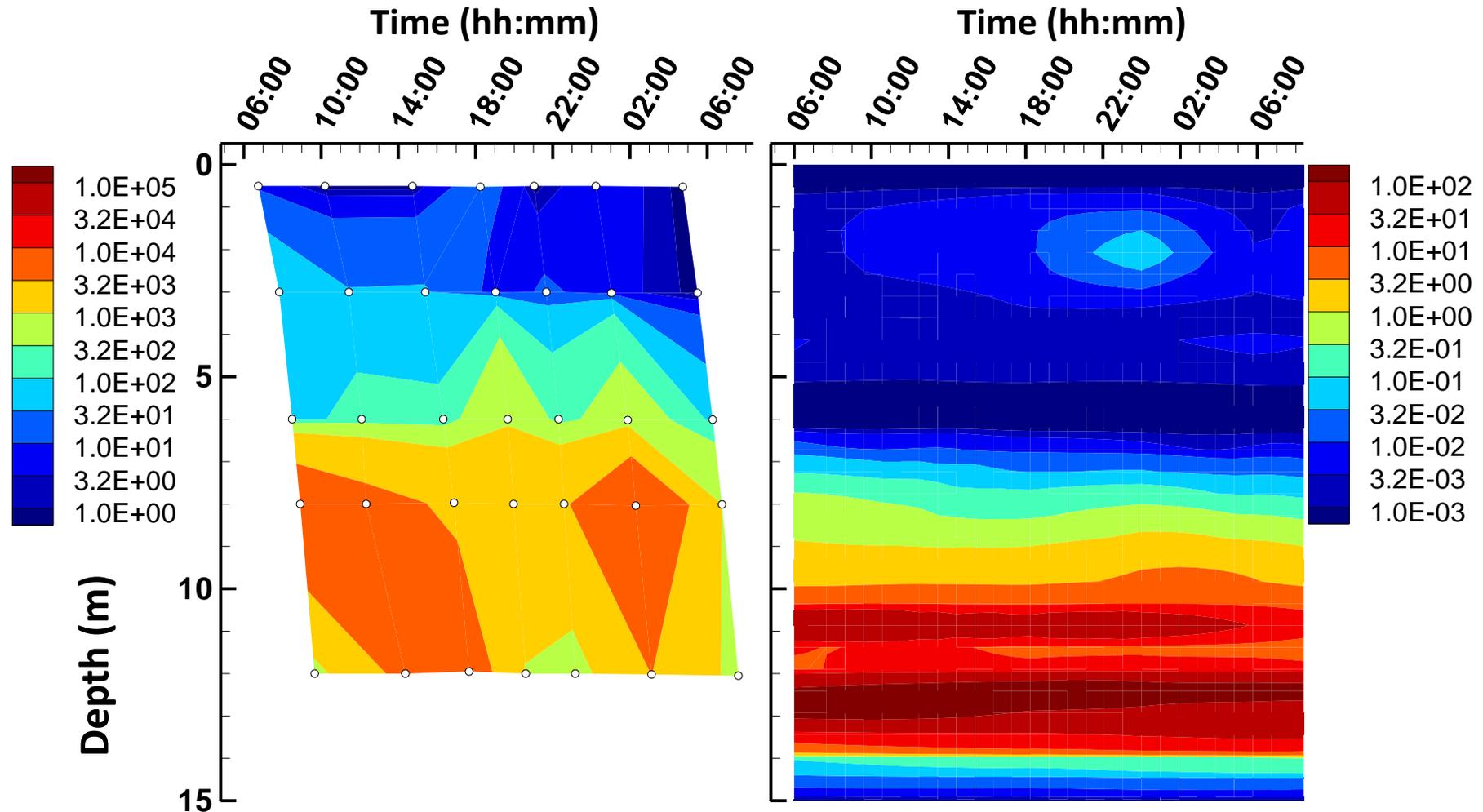
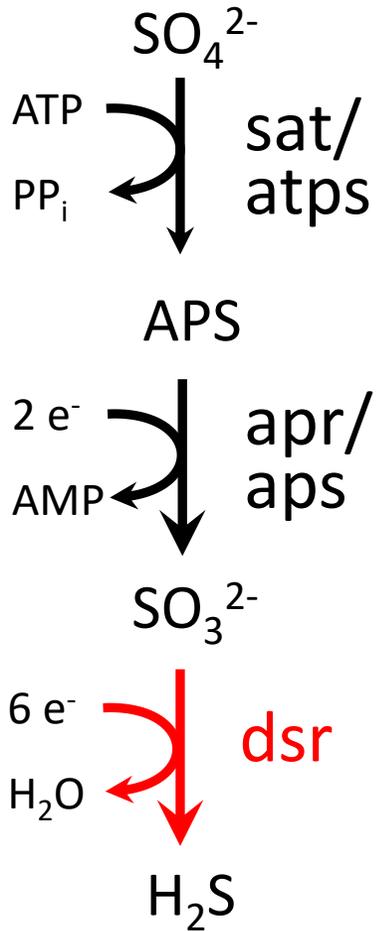
Short (0.25 d) vs Long (3 d) optimization: Phytoplankton



Breathing with Sulfate (sulfate reducing bacteria)

dsrA Transcript Abundance

$r_{1,SRB}$ ($\mu\text{M d}^{-1}$)*



dsr: dissimilatory sulfite reductase

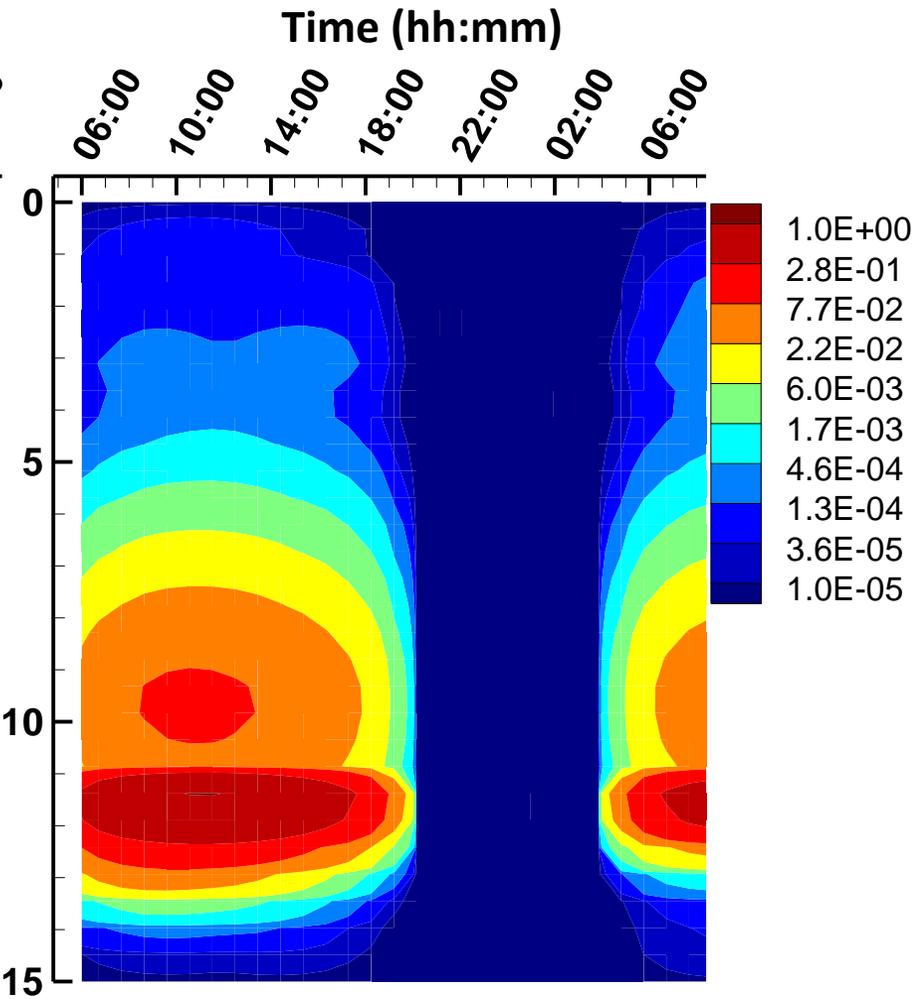
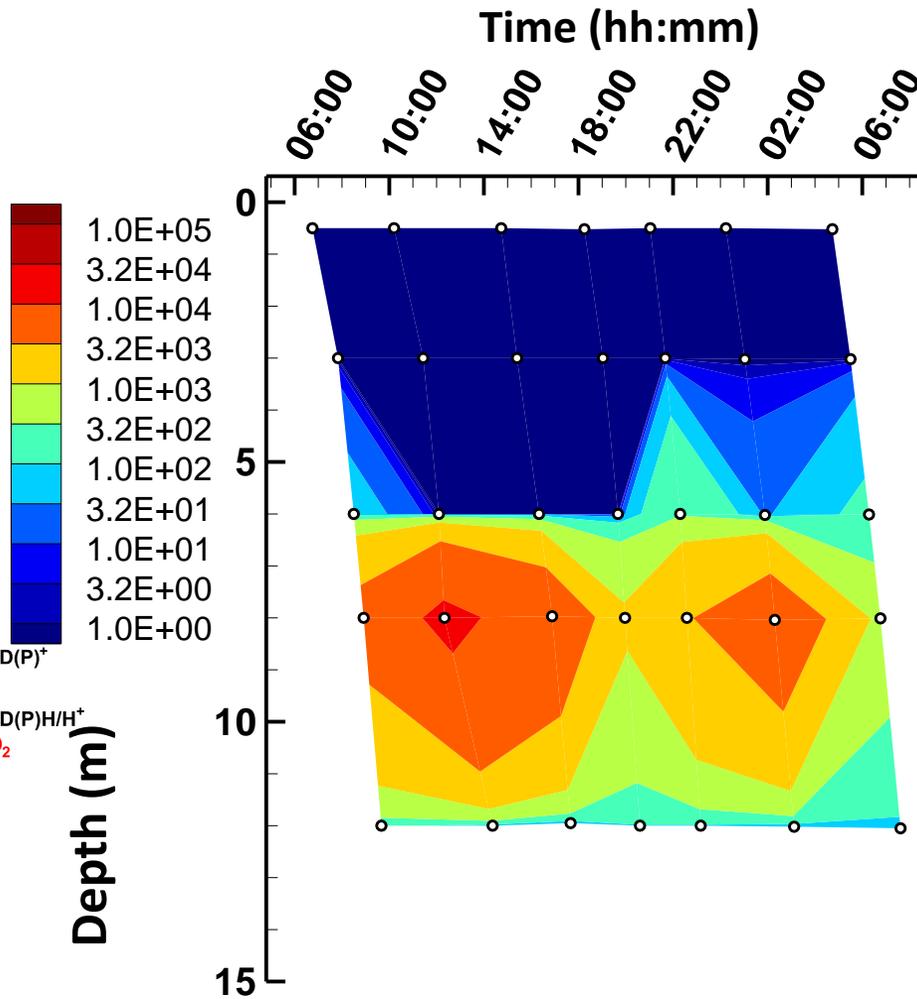
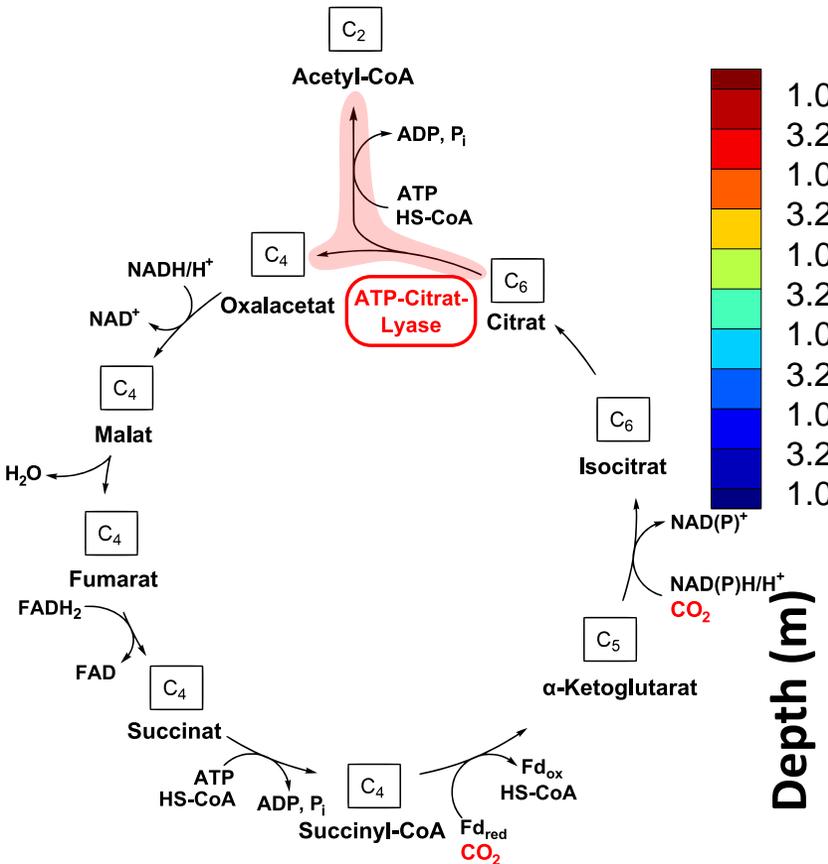
*(3 d Interval Optimization)

ATP-citrate lyase (anaerobic photosynthesis)

aclA Transcript Abundance

$r_{1,GSB} (\mu\text{M d}^{-1})^*$

Reverse TCA Cycle



*(3 d Interval Optimization)

Summary

- Maximizing entropy production (destruction of free energy) produces results that are similar to observations
 - Abiotic systems maximize instantaneous entropy production, while biotic system maximize entropy production over time using information.
 - Systems that coordinate information over space can increase global entropy production via coordination of function unless energy is degraded quickly abiotic (e.g., light)
- Model function not individuals (metabolic network)
- Replace parameters with control variables as much as possible
- The Control of PDE problem is computationally challenging
 - Faster solution approaches?
 - Different model formulation: Trait-based model optimized by MEP?

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